

# Stereoselective synthesis of (2*S*,3*S*,7*S*)-3,7-dimethylpentadec-2-yl acetate and propionate, the sex pheromones of pine sawflies

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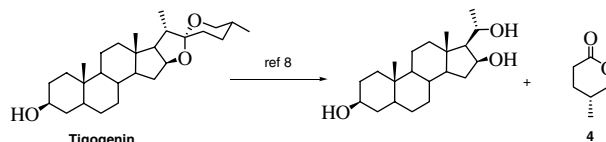
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**Abstract**—The stereoselective synthesis of (2*S*,3*S*,7*S*)-3,7-dimethylpentadec-2-yl acetate (**2**) and propionate (**3**) was accomplished by utilizing the cheap and easily available chiron (*R*)-4-methyl- $\delta$ -valerolactone (**4**). The key steps were chelation-controlled addition of Gilman reagent to chiral  $\beta$ -alkoxy aldehyde **12** and the Cu(I)-catalyzed coupling of Grignard reagent with bromoester **5** in the presence of NMP.

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Pine sawflies (Hymenoptera: *Diprionidae*) are widely distributed in the coniferous forests of Northern Europe, Asia and North America. Considering they are severe pests on conifers, the identification and synthesis of the pheromones of this and related species are essential for the development of selective monitoring and control of the populations of these insects. Because of the pioneering work of Coppel, Jewett and co-workers,<sup>1,2</sup> the major pheromone components of several species of pine sawflies were found to be esters of 3,7-dimethyl-2-pentadecanol **1**. Acetate **2** is the most active in the *Neodiprion* species, whereas propionate **3** is preferred by the *Diprion* species.<sup>2,3</sup> Further researches showed that the esters of (2*S*,3*S*,7*S*)-alcohol **1** were the most active for all *Neodiprion* species.<sup>4</sup> Up to date, a number of methods<sup>5–7</sup> have been developed for the syntheses of stereoisomers of compound **1**. Compared with the existing syntheses of those pheromones, the present route seems more practical and economical.

As a result of our group's efforts in resource chemistry (the organic synthetic chemistry based on the rational utilization of resource compounds), we can produce (*R*)-4-methyl- $\delta$ -valerolactone (**4**) in kilograms scales via



Scheme 1.

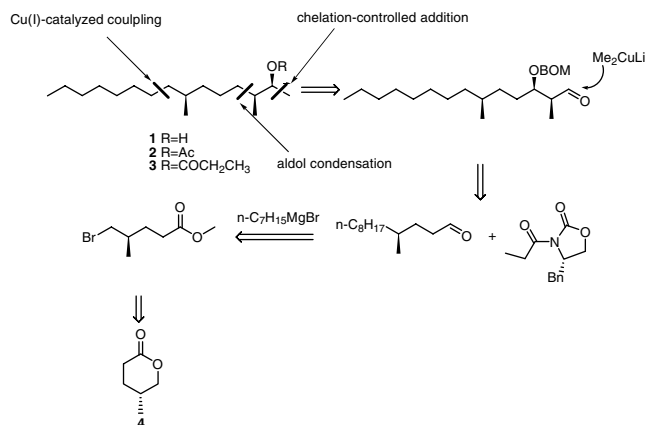
a green method (Scheme 1).<sup>8</sup> In connection with our work in using the cheap and easily available lactone **4** as a chiron for the asymmetric syntheses of natural products,<sup>9</sup> we report here a new route to the esters of (2*S*,3*S*,7*S*)-alcohol **1**.

Our retrosynthetic analysis is shown in Scheme 2. The chelation-controlled addition to aldehyde and the coupling of Grignard reagent with bromide were the key steps in this synthesis.

The asymmetric synthesis started with lactone **4**. On the treatment of HBr gas in MeOH,<sup>10</sup> lactone **4** was converted to bromoester **5**. In the presence of NMP, Cu(I)-catalyzed coupling<sup>11</sup> of Grignard reagent with bromoester was achieved in high chemoselectivity and yield. After reduction with LiAlH<sub>4</sub>, the alcohol was oxidized by Swern protocol<sup>12</sup> to provide aldehyde **7**. By a TiCl<sub>4</sub>-mediated aldol condensation<sup>13</sup> of the known *N*-propionyl-oxazolidinone<sup>14</sup> with the aldehyde, the desired compound **9** was achieved. Protection of the

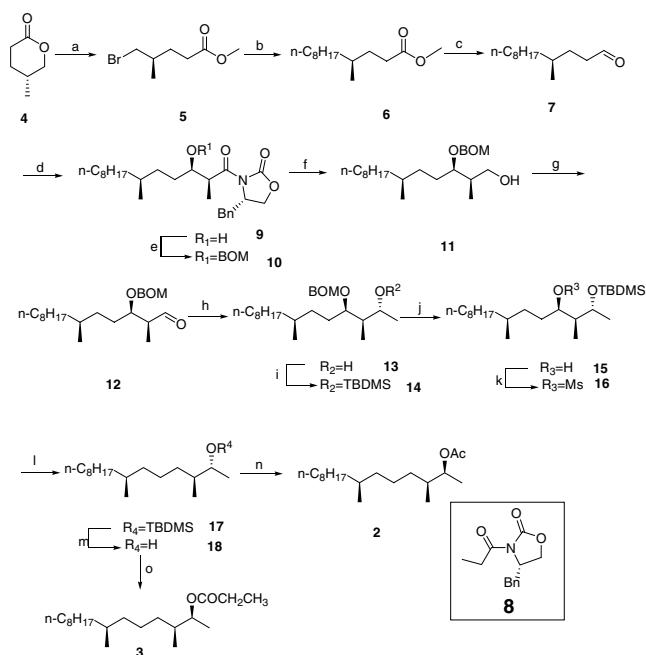
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Scheme 2. Retrosynthetic analysis of 1.

secondary alcohol group of resulting alcohol **9** with BOMCl, followed by reduction cleavage of the chiral auxiliary with NaBH<sub>4</sub> in a THF/H<sub>2</sub>O mixture,<sup>15</sup> provided primary alcohol **11**. After an IBX oxidation,<sup>16</sup> aldehyde **12** was treated with Gilman reagent to give the chelation-controlled adduct **13** as a single isomer.<sup>17</sup> TBDMS protection of **13**, followed by debenzyloxy-methylation, gave alcohol **15**. The secondary alcohol of compound **15** was removed via reduction of its mesylate **16** with NaBH<sub>4</sub> in NMP.<sup>18</sup> Deprotection of



**Scheme 3.** Reagents and conditions: (a) dry HBr, MeOH, rt, 56%; (b) 4 equiv NMP, 3% Li<sub>2</sub>CuCl<sub>4</sub>, CH<sub>3</sub>(CH<sub>2</sub>)<sub>5</sub>CH<sub>2</sub>MgBr, THF, rt, 92%; (c) (1) LiAlH<sub>4</sub>, Et<sub>2</sub>O, -78 °C, 90%; (2) DMSO, (COCl)<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>, -78 °C to 0 °C, 85%; (d) **8**, TiCl<sub>4</sub>, iPr<sub>2</sub>NEt, NMP, CH<sub>2</sub>Cl<sub>2</sub>, 0 °C, 82%; (e) BOMCl, iPr<sub>2</sub>NEt, TBAI, CH<sub>2</sub>Cl<sub>2</sub>, 0 °C to rt, 80%; (f) NaBH<sub>4</sub>, THF/H<sub>2</sub>O (v/v,1:2.5), 0 °C to rt, 87%; (g) IBX, DMSO, rt 88%; (h) Me<sub>2</sub>CuLi, Et<sub>2</sub>O, -78 °C, 76%; (i) TBDMSCl, imidazole, DMF, rt 95%; (j) H<sub>2</sub>, 10% Pd-C, EtOAc/MeOH (v/v,3:1), rt 90%; (k) MsCl, Et<sub>3</sub>N, CH<sub>2</sub>Cl<sub>2</sub>, 0 °C to rt, 96%; (l) NaBH<sub>4</sub>, HMPA, 80 °C, 78%; (m) HF (40%), CH<sub>3</sub>CN, rt, 94%; (n) DEAD, Ph<sub>3</sub>P, pyridine, AcOH, THF, -45 °C to 0 °C, 80%; (o) DEAD, Ph<sub>3</sub>P, pyridine, CH<sub>3</sub>CH<sub>2</sub>COOH, THF, -45 °C to 0 °C, 77%.

TBDMS followed by Mitsunobu reaction<sup>19</sup> afforded sex pheromones **2**<sup>20</sup> and **3**<sup>21</sup> (see Scheme 3).

In conclusion, starting from (*R*)-4-methyl- $\delta$ -valerolactone (**4**), which was regarded as industrial waste in the past, an efficient synthesis of the active enantiomers of the pheromones of pine sawflies has been achieved. The key steps of this synthesis include the highly diastereoselective addition to chiral  $\beta$ -alkoxy aldehyde, and the Cu(I)-catalyzed coupling of Grignard reagent with bromoester.

## References and notes

- Coppel, H. C.; Casida, J. E.; Dauterman, W. C. *An. Entomol. Soc. Am.* **1960**, *53*, 510.
- Jewett, D. M.; Matsumura, F.; Coppel, H. C. *Science* **1976**, *192*, 51.
- Jewett, D. M.; Matsumura, F.; Coppel, H. C. *J. Chem. Ecol.* **1978**, *4*, 277.
- (a) Tai, A.; Morimoto, N.; Yoshikawa, M.; Uehara, K.; Sugimura, T.; Kikukawa, T. *Agric. Biol. Chem.* **1990**, *54*, 1753; (b) Hedenström, E.; Högberg, H. E.; Wassgren, A. B.; Bergström, G.; Löfqvist, J.; Hansson, B. S.; Anderbrant, O. *Tetrahedron* **1992**, *48*, 3139.
- For racemic syntheses of **1**, see: (a) Ref. 2; (b) Magnusson, G. *Tetrahedron Lett.* **1977**, *31*, 2713; (c) Kocienski, P. J.; Ansell, J. M. *J. Org. Chem.* **1977**, *42*, 1102; (d) Place, P.; Roumestant, M. L.; Gore, J. *J. Org. Chem.* **1978**, *43*, 1001; (e) Baker, R.; Winton, P. M. *Tetrahedron Lett.* **1980**, *21*, 1175; (f) Kallmerten, J.; Balestra, M. *J. Org. Chem.* **1986**, *51*, 2855; (g) Gould, T. J.; Balesstra, M.; Wittman, M. D.; Gary, J. A.; Rossano, L. T.; Kallmerten, J. *J. Org. Chem.* **1987**, *52*, 3889; (h) Hedenström, E.; Högberg, H. E. *Tetrahedron* **1994**, *50*, 5225.
- For the syntheses of optically active **1**, see: (a) Magnusson, G. *Tetrahedron* **1978**, *34*, 1385; (b) Mori, K.; Tamada, S. *Tetrahedron Lett.* **1978**, *10*, 901; (c) Mori, K.; Tamada, S. *Tetrahedron* **1979**, *35*, 1279; (d) Byström, S.; Högberg, H. E.; Norin, T. *Tetrahedron* **1981**, *37*, 2249; (e) Kikukawa, T.; Imaida, M.; Tai, A. *Bull. Chem. Soc. Jpn.* **1984**, *57*, 1954; (f) Itoh, T.; Yonekawa, Y.; Sato, T.; Fujisawa, T. *Tetrahedron Lett.* **1986**, *27*, 5405; (g) Larchevêque, M.; Sanner, C. *Tetrahedron* **1988**, *44*, 6407; (h) Ref. 4a; (i) Högberg, H. E.; Wassgren, A. B.; Hjalmarsson, M.; Bergström, G.; Löfqvist, J.; Norin, T. *Tetrahedron* **1990**, *46*, 3007; (j) Huang, P.-Q.; Lan, H.-Q.; Zheng, X.; Ruan, Y.-P. *J. Org. Chem.* **2004**, *69*, 3964; (k) Bekish, A. V.; Prokhorevich, K. N.; Kulinkovich, O. G. *Eur. J. Org. Chem.* **2006**, 5069.
- For some syntheses of homologous and diastereomers of **1**, see: (a) Kikukawa, T.; Imaida, M.; Tai, A. *Chem. Lett.* **1982**, 1799; (b) Tai, A.; Sugimura, T.; Kikukawa, T.; Naito, C.; Nishimoto, Y.; Morimoto, N. *Biosci., Biotechnol., Biochem.* **1992**, *56*, 1711; (c) Ref. 4b; (d) Tai, A.; Higashiura, Y.; Kakizaiki, M.; Naito, T.; Tanaka, K.; Fujita, M.; Sugimura, T.; Hara, H.; Hayashi, N. *Biosci., Biotechnol., Biochem.* **1998**, *62*, 607; (e) Moreira, J. A.; Corrêa, A. G. *J. Braz. Chem. Soc.* **2000**, *11*, 614; (f) Ebert, S.; Krause, N. *Eur. J. Org. Chem.* **2001**, 3831; (g) Hedenström, E.; Edlund, H.; Lund, S.; Abersten, M.; Persson, D. *J. Chem. Soc., Perkin Trans. 1* **2002**, 1810; (h) Tai, A.; Tanaka, K.; Fujita, M.; Sugimura, T.; Higashiura, Y.; Kasashi, M.; Hara, H.; Naito, T. *Bull. Chem. Soc. Jpn.* **2002**, *75*, 111.
- Tian, W. S. CN1061985C.

9. Tian, W. S.; Ding, K.; Huang, Y. CN 200410093260.4.
10. Paquette, L. A.; Dahnke, K.; Doyon, J.; He, W.; Wyant, K.; Friedrich, D. *J. Org. Chem.* **1991**, *56*, 6199.
11. Cahiez, G.; Chaboche, C.; Jézéquel, M. *Tetrahedron* **2000**, *56*, 2733.
12. Mancuso, A. J.; Swern, D. *Synthesis* **1981**, 165.
13. Crimmins, M. T.; She, J. *Synlett* **2004**, 1371.
14. (a) Gage, J. R.; Evans, D. A. *Org. Synth.* **1989**, *68*, 77; (b) Evans, D. A.; Gage, J. R.; Leighton, J. L. *J. Am. Chem. Soc.* **1992**, *114*, 9434.
15. Prashad, M.; Kim, H.-Y.; Lu, Y.; Har, D.; Repic, O.; Blacklock, T. J.; Giannousis, P. *J. Org. Chem.* **1999**, *64*, 1750.
16. Frigerio, M.; Santagostino, M. *Tetrahedron Lett.* **1994**, *35*, 8019.
17. Still, W. C.; Schneider, J. A. *Tetrahedron Lett.* **1980**, *21*, 1035.
18. (a) Torisawa, Y.; Nishi, T.; Minamikawa, J.-i. *Bioorg. Med. Chem. Lett.* **2001**, *11*, 2787; (b) Torisawa, Y.; Nishi, T.; Minamikawa, J.-i. *Bioorg. Med. Chem.* **2002**, *10*, 2583.
19. Mitsunobu, O. *Synthesis* **1981**, 1.
20. (2*S*,3*S*,7*S*)-3,7-Dimethylpentadec-2-yl acetate (**2**):  $[\alpha]_{\text{D}}^{25}$   $-5.6$  (*c* 0.18, hexane); IR(film): 2963, 2928, 2857, 1741, 1465, 1374, 1261 1246, 1098, 1020, 803  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 300 MHz)  $\delta$ : 4.70–4.83 (m, 1H), 2.03 (s, 3H), 0.95–1.60 (m, 22H), 1.16 (d,  $J = 6.3$  Hz, 3H), 0.76–0.84 (m, 9H);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 75 MHz)  $\delta$ : 170.8, 74.137.6, 37.3, 37.0, 32.8, 32.8, 31.9, 30.0, 29.7, 29.4, 27.0, 24.5, 22.7, 21.3, 19.7, 17.0, 14.8, 14.1; MS EI  $m/z$  (%): 239 ( $\text{M}^+ - 59$ , 1), 154(1), 140(3), 125(4), 111(3), 97(5), 87(29), 72(35), 57(23), 43(100).
21. (2*S*,3*S*,7*S*)-3,7-Dimethylpentadec-2-yl propionate (**3**):  $[\alpha]_{\text{D}}^{25}$   $-5.8$  (*c* 0.35, hexane); IR(film): 2962, 2928, 2856, 1739, 1464, 1378, 1261 1193, 1084, 1019, 805  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 300 MHz)  $\delta$ : 4.78–4.87 (m, 1H), 2.30 (q,  $J = 7.5$  Hz, 2H), 1.16 (d,  $J = 6.6$  Hz, 3H), 1.14 (t,  $J = 7.6$  Hz, 3H), 0.99–1.60 (m, 22H), 0.93–0.76 (m, 9H);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 75 MHz)  $\delta$ : 174.1, 73.8, 37.7, 37.3, 37.0, 32.9, 32.7, 31.9, 30.0, 29.7, 29.4, 28.0, 27.1, 24.5, 22.7, 19.7, 17.0, 14.8, 14.1, 9.3; MS EI  $m/z$  (%): 238 ( $\text{M}^+ - 74$ , 2), 155(1), 139(3), 125(4), 111(3), 101(20), 86(21), 69(15), 57(100).